SYNTHESIS OF p-THIOCYANOARYLDIOXOPHOSPHAZANES

N. N. Mel'nikov, K. D. Shvetsova-Shilovskaya, and L. V. Nifant'eva

Khimiya Geterotsiklicheskikh Soedinenii, Vol. 2, No. 3, pp. 465-466, 1966

p-Thiocyanophenyldioxophosphazanes are synthesized by reacting alkylphosphonic dichlorides with p-thiocyanoanilines at room temperature.

Phosphonic dichlorides react with anilines and other aromatic amines to give the corresponding phosphodianilides [1]. When the latter are heated under drastic conditions, they can split off amine, and undergo conversion to polyamides or heterocyclic compounds containing 4-membered rings, aryldioxophosphazanes [2]. The literature also describes in-direct preparation of these 4-membered ring compounds by reacting phosphonic dichlorides with amines, with prolonged heating of the reaction mixture to $100^{\circ}-120^{\circ}$ C [3].

During an investigation of the phosphorylation of substituted p-thiocyanoanilines, we studied the reaction of 4thiocyanoaniline, 3-chloro-4-thiocyanoaniline, and 3-methyl-4-thiocyanoaniline with methyl- and phenylphosphonic dichlorides, and showed that the reaction is peculiar, the corresponding heterocyclic compounds being formed even at room temperature:

$$CH_{3} - PCI_{2} + H_{2}N - C_{6}H_{4} - SCN - p \xrightarrow{\text{pyridine}} CH_{3} - \frac{V_{6}H_{4}SCN - p}{V_{N}} \xrightarrow{P} - CH_{3}$$

The structures of the compounds obtained follows from the elementary data, molecular weight, and IR spectra. Formation of cyclic compounds containing phosphorus, and not of phosphonic dianilides is shown by the IR spectrum not containing bands characteristic of the NH fragment. It should be mentioned that the thiocyano group is unaffected by the reaction, as the IR spectrum has a band at 2160 cm^{-1} [4].

p-Thiocyanoaryldioxophosphazanes	R



					Fc	ound,	%	Calc	ulated	1, %	90
No.	R	Ar	Mp °C	Formula	CI	N	s	C1	N	s	Yield,
1	CH₃	C ₆ H ₄ SCN-p	140—141	$C_{16}H_{14}N_4O_2P_2S_2$		13.08 12.90			13.35	15.25	10
2	СНз	C ₆ H ₃ (Cl) SCN-3,4	147—148	$C_{16}H_{12}CI_2N_4O_2P_2S_2$		11.35 11.03		14.55	11.45	13.1	12
3	CH₃	C ₆ H ₃ (CH ₃)SCN-3,4	151—152	$C_{18}H_{18}N_4O_2P_2S_2$		$12.43 \\ 12.35$			12.5	14.3	7
4	C ₆ H ₅	C ₆ H ₄ SCN- <i>p</i>	123—124	$C_{26}H_{18}N_4O_2P_2S_2$		$10.12 \\ 10.41$	11.53 11.55		10.3	11.75	13
5	C_6H_5	C ₆ H ₃ (Cl) SCN-3,4	154—155	$\mathrm{C}_{26}\mathrm{H}_{16}\mathrm{CI}_{2}\mathrm{N}_{4}\mathrm{O}_{2}\mathrm{P}_{2}\mathrm{S}_{2}$	11.05 11.05		9.9 9.9	11.6	9.15	10.04	15
6	C ₆ H₅	$C_6H_3(CH_3)SCN-3,4$	160—161	$C_{28}H_{22}N_4O_2P_2S_2$		9.72 10.03	11.32 11.52		9.8	11.18	15

Experimental

<u>1,3-Dimethyl-3, 4-di-p-thiocyanophenyldioxophosphazane (I)</u>. 1.3 g (0.1 mole) methylphosphonic dichlorides in 10 ml dichloroethane was added dropwise, at 20° C, to 1.5 g (0.1 mole) p-thiocyanoaniline in 50 ml dry dichloroethane and 2.1 ml dry pyridine. The reaction mixture was left overnight, the precipitated pyridine hydrochloride filtered off, the reaction products washed a few times with cold water, the solvent vacuum-distilled off, and the residue, a viscous oil, treated with EtOH, when crystals separated, mp 140°-141° C, yield 10%. Found: N 13.08, 12.90; S 14.90, 14.85%; M 472, (isothermal distillation method). Calculated for $C_{16}H_{14}N_4O_2P_2S_2$: N 13.35; S 15.25%; M 489.

The other compounds for which data are given in the table were prepared similarly.

REFERENCES

- 1. G. M. Kosolapoff, Organophosphorus Compounds, N.Y., 279, 1950.
- 2. V. Gutman, D. E. Hagen, and K. Utvary, Mon., 93, 747, 1962.
- 3. H. Binder and R. Heinle, West German Republic patent no. 1 083 818; C. A., 55, 17543, 1961.
- 4. L. J. Bellamy, The Infra-Red Spectra of Complex Molecules [Russian translation], IL, Moscow, 1957.

28 June 1965

All-Union Scientific Research Institute of Chemical Plant Protectants, Moscow